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Determination Of Ph Of Aqueous Extracts Part 1

Cold Extraction

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the concept of expressing acidity as the negative logarithm of the hydrogen ion concentration was defined and termed pH in the beginning of the 20th century the general usefulness of the pH concept for life science was recognized and later gained importance to analytical research reports on results of pH measurements from living skin established the term acid mantle the skin's own protective shield that maintains a naturally acid pH it is invisible to the eye but crucial to the overall wellbeing of skin chronic alkalization can throw this acid mantle out of balance leading to inflammation dermatitis and atopic skin diseases it is therefore no surprise that skin pH shifts have been observed in various skin pathologies

it is also obvious that the ph in topically applied preparations may play an important role optimal ph and buffer capacity within topical preparations not only support stability of active ingredients and auxiliary materials but may also increase absorption of the non ionized species of an acidic or a basic active ingredient they may even open up opportunities to modify and correct skin ph and hence accelerate barrier recovery and maintain or enhance barrier integrity further efforts are needed to standardize and improve ph measurements in biological media or pharmaceutical cosmetic vehicles to increase and ensure quality comparability and relevance of research data in this volume we present a unique collection of papers that address past present and future issues of the ph of healthy and diseased skin it is hoped that this collection will foster future efforts in clinical and experimental skin research

whether ph is being used to test a sample against a legal requirement or specification as part of an analytical method for monitoring and controlling a reaction as a process control in the chemical industry or for the environmental monitoring of waste and effluents it is important that all ph measurements are carried out in a logical and consistent manner paying careful attention to experimental procedures in order to obtain reliable results this guide provides scientists with the knowledge of how to do just that first by outlining the principles of ph measurement and buffer solutions ph meters and electrodes are then discussed including selection criteria and the care of electrodes finally sections on making ph measurements and uncertainty are followed by a set of practical exercises measurement of ph is one of the practical laboratory skills training guides a series that aims to make achieving best practice easy these invaluable manuals will enable both experienced and inexperienced staff to get the essential basics of any experiment right simply by following the clear and easy to use instructions provided the guides are written by experienced scientists and include minimal theory plenty of practical exercises in order to assess competence and trouble shooting information other titles are measurement of mass measurement of volume high performance liquid chromatography and gas chromatography

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